Synthesis and Characterization of Novel Dimethylgallium Complexes

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Abstract: Two novel dimethylgallium complexes were synthesized and their structures were elucidated by elemental analysis, ¹H NMR, IR and MS spectra.

Keywords: Novel dimethylgallium complexes, synthesis, characterization.

Recently, great attention has been paid to Group IIIA organometallic compounds owing to their wide range of uses in the areas of ceramic semi-conductor materials and diagnosis of diseases¹. However, most of the relevant studies are limited to monoaminomethylated phenols as ligands².



In this paper, we report the synthesis of compounds (1a and 2a) as a new ligand and their dimethylgallium complexes (1b and 2b) (Scheme 1). General procedures for preparation of 1a, 2a and 1b, 2b were accorded with ref. 2, 3. All new compounds were characterized by elemental analysis, ¹H NMR, IR and MS spectra (see Table 1).

The complexes are much more stable than $GaMe_3$, and decompose slowly in air. The elemental analysis values of $GaMe_3$ complexes match with calculated (see **Table 1**). A moderate absorption near 530 cm⁻¹ in IR spectrum shows the Ga-N stretching vibration

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frequency². In the ¹H NMR spectrum, there is a single peak at -0.29 or -0.49 ppm, which is the characteristic signals of protons on the methyl group connecting to Ga. In the MS showed two isotopes Ga⁶⁹ and Ga⁷¹ signals with the abundance ratio 3:2, the *m/z* of [M]⁺, [M-CH₃]⁺, [Ga]⁺, [Ga (CH₃)₂]⁺ and other fragments can confirm the structures of the complexes further.

Compd.	mp	IR, (KBr,	¹ H NMR	MS(EI)	Elementa
	(°C)	cm^{-1})	(300MHz, CDCl ₃ ,	m/z(intensity %)	l Analysis
			δ ppm)		C, H, N
					(%)
		3020, 2950,	2.56 (s, 8H, NCH ₂),	308.1 (M ⁺ , 12.53),	Calcd:
		1510, 1475,	3.64 (s, 8H, OCH ₂),	221.1 (100.00),	62.32,
		1445, 1375,	3.75 (s, 4H, PhCH ₂ N),	162.1 (6.15),	7.85, 9.08
1a	203-5	1320, 1250,	6.50 (s, 2H, ArH),	136.1 (15.23),	Found:
		1185, 1100,	9.70-10.30 (br.s, 1H,	86.1 (56.68),	62.26,
		980, 905,	PhOH)	57.0 (65.62)	7.71, 9.19
		860, 750			
		3020, 2950,	-0.29 (s, 12H, GaCH ₃),	508.0 (M ⁺ , 3.67),	Calcd:
		1475, 1440,	2.54-2.61 (t, 4H,	506.0 (11.02),	47.46,
		1345, 1310,	NCH ₂),	503.9 (6.71),	6.77, 5.54
1b	>250	1250, 1205,	3.10-3.14 (t, 4H,	493.5 (0.96), 491.3 (2.76),	Found:
	deco	1095, 960,	NCH ₂),	488.9 (2.46), 404.9 (3.13),	47.50,
	mp.	900, 860,	3.61-3.69 (t, 8H,	304.0 (7.61), 101.0(38.32),	6.90, 5.45
		770, 580,	OCH ₂),	99.0 (57.66), 70.9 (17.04),	
		530	3.90-3.94 (d, 4H,	68.9 (27.08)	
			CH ₂ Ph),		
			6.56 (s, 2H, ArH)		
		3300,3020,	δ (DMSO):	312.0 (M ⁺ , 0.27),	Calcd:
		2950,1600,	-0.49 (s, 6H, GaCH ₃),	310.0 (0.48),	57.91,
2b	>150	1480,1350,	6.76-7.57 (m, 9H, ArH),	297.0 (7.66), 295.0(13.33),	5.51, 9.00
	deco	1260,1118,	8.16 (s, 1H, CH=N),	211.1 (21.16), 101.0(1.73),	Found:
	mp.	990,740,	10.41 (br.s, 1H, NH)	99.0 (2.41), 70.9 (5.27),	58.08,
		680,530		68.9 (8.11)	5.63, 9.01

Table1 The spectral analytical and physical data of new compounds

The research of the complex analogues is in progress in our laboratory.

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